

# Radioactivity

17.3

Assume a radioactive material  $Q(t)$  decays at rate proportional to the amount present at any particular time  $\Rightarrow$

$$\frac{dQ}{dt} = -r Q, \text{ where } -r < 0 \text{ is the decay rate}$$

This DE has the form of (A) "1<sup>st</sup> order linear DE with constant coefficient"  $\Rightarrow$  so we can use (A\*) to find the quantity  $Q(t)$  available at time  $t \Rightarrow$

$$Q(t) = \frac{b}{a} + \left(Q_0 - \frac{b}{a}\right) e^{at} \quad (A^*) \text{ where}$$

$Q_0 = Q(0)$  is the initial amount of this material

$$a = -r \text{ and } b = 0 \text{ so } \frac{b}{a} = 0$$

$$Q(t) = Q_0 e^{-rt} \text{ is the solution of this problem}$$

**Half-life time  $\tau$** : is the time takes the material to decay by the half of its initial

$Q(\tau) = \frac{1}{2} Q_0$	$-r\tau = \ln \frac{1}{2}$	Note that $-r < 0 \Rightarrow r > 0 \Rightarrow \tau > 0$
$Q_0 e^{-r\tau} = \frac{1}{2} Q_0$	$-r\tau = -\ln 2$	
$e^{-r\tau} = \frac{1}{2}$	$\tau = \frac{\ln 2}{r}$	



Exp (Q14 - section 1.2)

Radium-226 has half-life of 1620 years.  
Find the time period during which a given amount of this material is reduced by one-quarter.

$$T = \frac{\ln 2}{r} \Rightarrow r = \frac{\ln 2}{T} = \frac{\ln 2}{1620}$$

$Q(t) = Q_0 e^{-rt}$  is the amount available at time  $t$

we need to find the time  $t^*$  s.t

$$Q(t^*) = \frac{3}{4} Q_0 \quad \text{since } \frac{1}{4} \text{ is reduced}$$

$$Q_0 e^{-rt^*} = \frac{3}{4} Q_0$$

$$e^{-rt^*} = \frac{3}{4}$$

$$-rt^* = \ln \frac{3}{4}$$

$$t^* = \frac{\ln \frac{3}{4}}{-r} = \frac{\ln(0.75)}{-\frac{\ln 2}{1620}} = -1620 \frac{\ln(0.75)}{\ln 2}$$

$$\approx 672.4 \text{ years}$$