

Phys111 Report

8.5/10

Experiment #1: Density of a Metal and Distance between Atoms

Name:	
Partner:	u
Section:	3
Date:	6/4/2022

(1) Abstract:

Aim of the experiment:

is to find the density of a metal as well as ~~find~~ the distance between these atoms.

The main results are:

- The density of the metal block is $\rho = 7,40 \pm 0,04 \text{ g/cm}^3$
- The distance between atoms is $a = 2,32 \text{ \AA} = 2,32 \times 10^{-8} \text{ cm}$.

(2) Data:

Block #: S32

Mass (M) = ~~87,5~~ $\pm 0,1 \text{ gm}$

	1.	2.	3.	4.	5.	6.
L (cm)	4,000	3,980	3,990	4,010	4,005	4,000
W (cm)	1,900	1,915	1,920	1,930	1,905	1,925
T (cm)	1,540	1,545	1,546	1,547	1,549	1,543

Accepted
6/4/2022

(3) Calculations:

$\bar{L} = 3,4975$ cm	$\sigma_s(L) = 0,01083$	$\Delta\bar{L} = 0,004$ cm
$\bar{W} = 1,915$ cm	$\sigma_s(W) = 0,01158$	$\Delta\bar{W} = 0,005$ cm
$\bar{T} = 1,545$ cm	$\sigma_s(T) = 0,0031672$	$\Delta\bar{T} = 0,0013$ cm

$$V = \bar{L} \times \bar{W} \times \bar{T} = 3,4975 \times 1,915 \times 1,545 = 11,8273 \text{ cm}^3 \rightarrow 11,83 \text{ cm}^3$$

$$\frac{\Delta V}{V} = \frac{\Delta\bar{L}}{\bar{L}} + \frac{\Delta\bar{W}}{\bar{W}} + \frac{\Delta\bar{T}}{\bar{T}} = \frac{\Delta V}{V} = \frac{0,004}{3,497} + \frac{0,005}{1,915} + \frac{0,0013}{1,545} = 4,4531 \times 10^{-3}$$

$$\Delta V = 0,05 \text{ cm}^3$$

$$\rho = \frac{M}{V} = \frac{87,5}{11,8273} = 7,39813 \text{ g/cm}^3 \rightarrow 7,40 \text{ g/cm}^3$$

$$\frac{\Delta\rho}{\rho} = \frac{\Delta M}{M} + \frac{\Delta V}{V} = \frac{0,1}{87,5} + \frac{0,05}{11,8273} = 5,37036 \times 10^{-3}$$

$$\Delta\rho = 0,03973 \rightarrow 0,04 \text{ g/cm}^3$$

$$a = \sqrt[3]{\frac{A_w}{N_{AP}}} = \sqrt[3]{\frac{55,845}{6,023 \times 10^{23} \times 7,4}} = 2,322 \times 10^{-8} \text{ cm} = 2 \text{ \AA} = 2,32 \text{ \AA}$$

(4) Results:

- The density of the metal block is $\rho = 7,40 \pm 0,04 \text{ g/cm}^3$.
- The distance between atoms is $a = 2,32 \text{ \AA}$.

(5) Conclusions:

We measured the value of iron's density and we got $7,40 \text{ g/cm}^3$ with an error of about $0,04 \text{ g/cm}^3$. But the true value of iron's density is $7,88 \text{ g/cm}^3$. So when we use discrepancy test, we found $(|P_{me} - P_{re}| \leq 2 \times 0,08) \rightarrow (0,46 \leq 0,08)$ an error, I think that this error reasons are that our piece wasn't a completely rectangle, maybe we use the micrometer in a wrong way by compressed it a lot, and maybe my partner wasn't good at reading the distance information. However, we measured (L, W, T) six times, to reduce the random error.