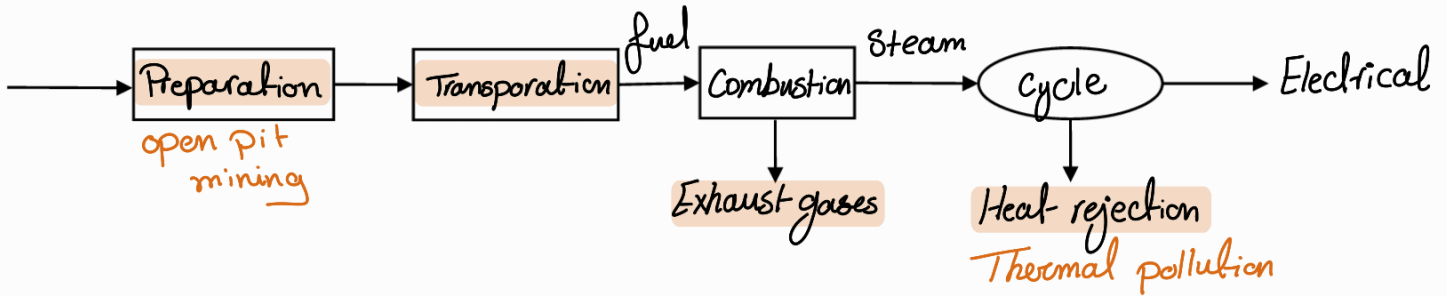
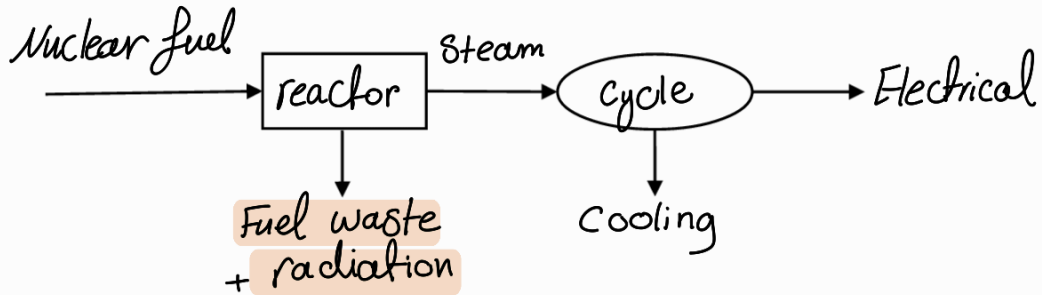


Chapter 5: Energy Conversion

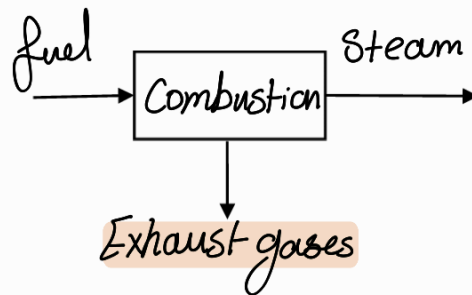
- Steam power plant pollution sources



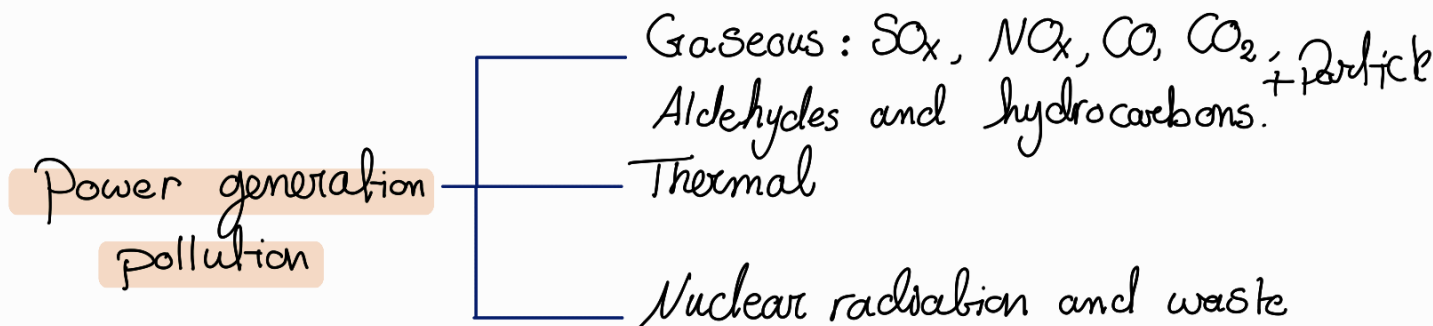
- Nuclear power plant pollution sources



- Gas turbine pollution sources



■ : Pollution sources



Gasolus

▷ SO_x gas \implies Slide 10 for details

- SO_2 and SO_3

- Toxic and cause irritation to mucus membrane so it harms people

- Causes acid rain: $SO_3 + H_2O \longrightarrow H_2SO_4$ which harms vegetation (It has acidity lower than $pH = 5.6$)

▷ CO_2 gas

- Green house gas

Allows short wave sun radiation \rightarrow Absorbs and bounces long wave radiation \rightarrow Prevent earth cooling \rightarrow Earth T increases

▷ CO gas \implies Slide 13 for details

- Colorless odorless gas that results from incomplete combustion

- CO sources: Automobile

Industrial processes

Residential processes

- CO effects: Harmful for humans

1. Reduces O_2 delivery to tissues and organs

2. Impair visual perception

3. Can be life threatening

▷ NO_x gas

- NO , NO_2 and other minor nitrogen oxides

- Combustion product

- NO_x sources: Automobile

Industrial processes

Residential processes

- NO_x effects:
 1. Causes health problems
 2. Creates smog
 3. Acid rain
 4. Reduces visibility in urban areas

□ Smog

CFC: chloro fluoro carbons

- Smoke and fog

← NO_2

- Composed from: NO_x , SO_x , O_3 , Smoke, particulates and other less visible pollutants (CO, CFC and radioactive sources)
 - + Photochemical oxidation + Sunlight

□ Particulates PM

- Effects:
 1. Harmful to humans (Breathing problems)
 2. Environmental (Visibility impairment, atmospheric deposition, Aesthetic damage and blanket sun radiation)
- Sources:
 - Industrial processes
 - Transportation
 - Residential
 - Dust from roads, construction and agricultural emissions

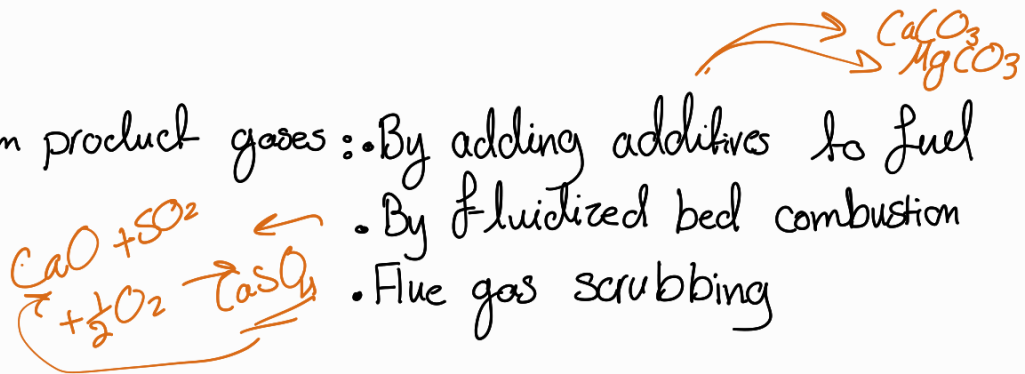
Air pollution Standards

1. Emission rate standard (Concentration of pollutants emitted)
2. Ambient concentration standard (Concentration of pollutants at ground level)

▷ SO_2 control by: 90% reduction H
70% ~ L

1. Burning low sulfur fuel → Expensive
2. Reducing sulfur content of fuel before combustion → Costly

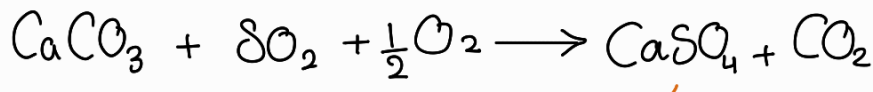
3. Removing SO₂ from product gases : • By adding additives to fuel
 (Slide 29) • By fluidized bed combustion
 • Flue gas scrubbing



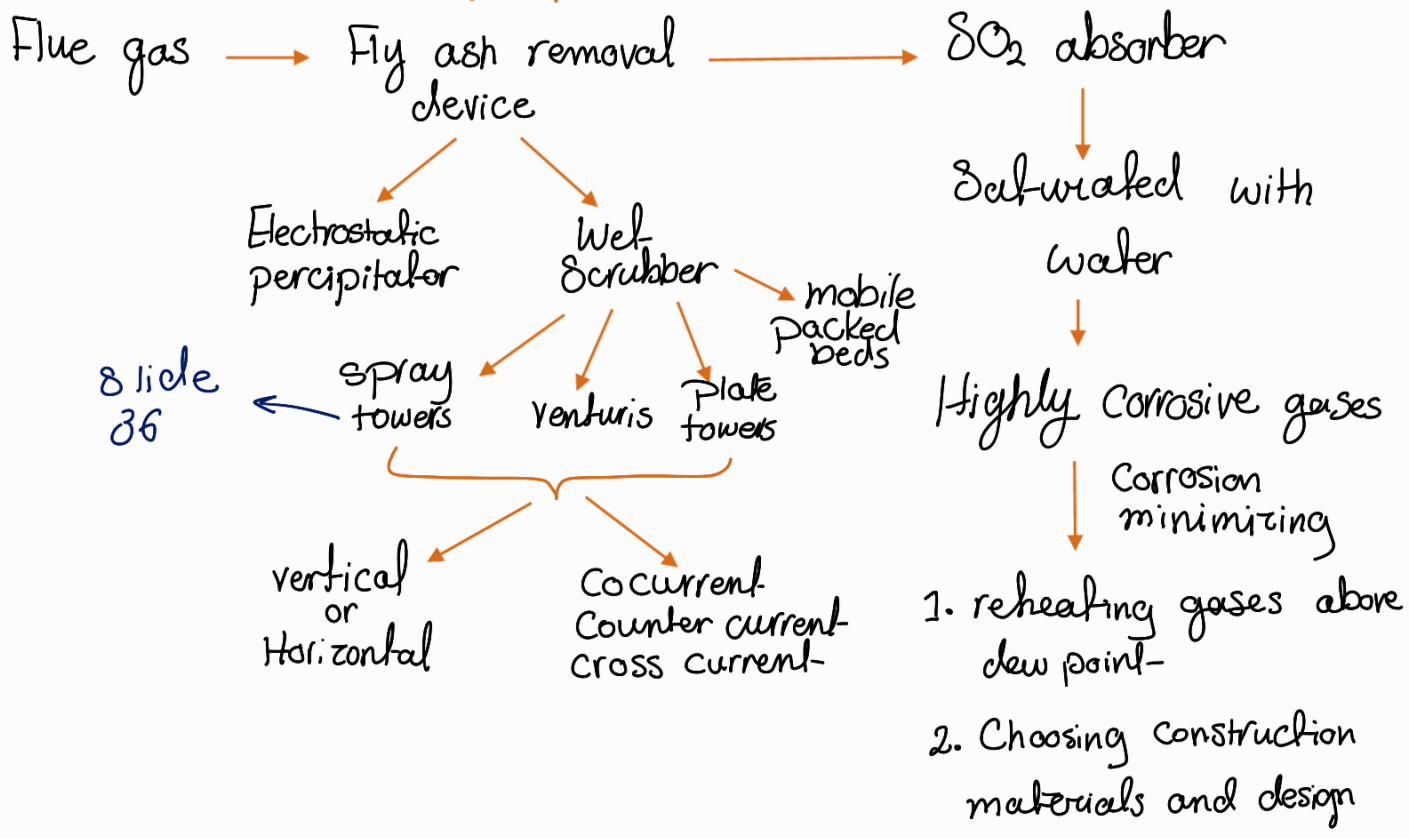
- Flue gas scrubbing : Scrubbing the flue gases
 1. Wet Scrubbing (FGD)
 2. Dry Scrubbing

○ Wet- Scrubbing

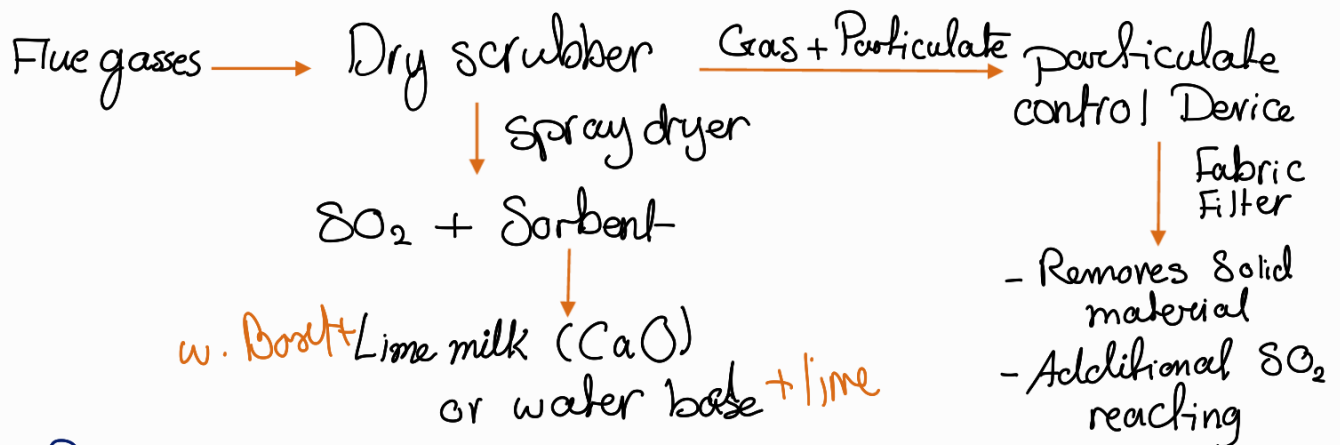
FGD reaction



Thrown away or re used to produce H₂SO₄



- Dry scrubbing
 - Dry injection or spray drying



For Process: Slide 38

▷ NO_x Control

- * Thermal NO: NO resulting from combustion air
- \rightarrow Higher flame temperature produces larger amounts of NO
- \rightarrow Flame temperature is influenced by:
 1. Rate of furnace cooling
 2. Combustion air temperature
 3. Excess air
- * Fuel NO_x
- \rightarrow NO_x from fuel which is a function of nitrogen in fuel
- * NO_x control method
 1. Flue gas recycling
 2. Two stage combustion & off stoichiometric firing
 3. Increasing the cooling rate

* Control techniques

1. Low-excess air firing
2. Staged combustion
3. Flue gas recirculation
4. Low NO_x burners
5. Re-burning

○ Re-burning

Taking a 10-20% of fuel and inject it to create a rich fuel in the reburn zone \rightarrow reduces NO_x

- Two-stage combustion or off-stoichiometric firing: some of burners are operated fuel rich and others are air rich \rightarrow reduces NO_x by a 5-6 factor

▷ Particulate control

- Effects: 1. Cause respiratory problems

2. Eye irritation

3. Enhance smog formation

- Types: Dust / Fume / Mist / Smoke / Spray

100	0.03-0.3	0.5-3	0.05-1	10-100
μm	μm	μm	μm	μm

- Control methods

1. Electrostatic precipitators

2. Bag filters

3. Scrubbers (Wet)

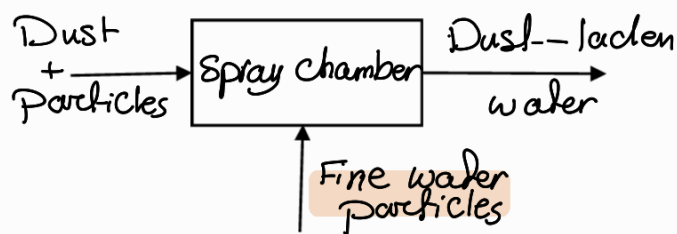
4. Mechanical dust collection (Cyclones)

○ Cyclones Slide 54

- Removes coarse particles
- Separates particles from air by exploiting their inertia
- Used before fine particles control devices
- Advantages:
 1. Simple design
 2. Small floor area
 3. low pressure drop
 4. High particle loading rates

○ Wet scrubber

- Washes dust and particles



○ Electrostatic precipitators (EPS) Slide 61

- Uses electrostatic charge to separate particles
- Consists of wires and plates with high voltage difference

Grounded

○ Bag filter Slide 65?

- Used for:
 - Expensive material (high collect-efficiency needed)
 - Fine sized particulate

Thermal pollution

- Thermal discharge index (TDI) :

$$\text{TDI} = \frac{\text{Thermal power discharged to environment } \text{MW}_{th}}{\text{Electric power output } \text{MWe}}$$
$$= \frac{1 - \eta_{Th}}{\eta_{Th}}$$

- Heat is rejected by condenser cooling system and exhaust-gases + heat losses in cycle

- Effects: Increases T of water bodies

Fresh water \rightarrow Aquatic life is harmed

Salt water \rightarrow Marine life is enhanced

- Limitations are on cooling systems \rightarrow Cooling towers are used

- EPA standards are 2.8°C increase in temperature for once-through cooling system

Nuclear waste

- Radioactive waste: Number of radio isotopes which are unstable configurations of elements that decay, emitting ionizing radiation.

- Sources: 1. Nuclear fuel cycle: In All processes of nuclear fuel cycle, from front end to back end

2. Medical and industrial wastes

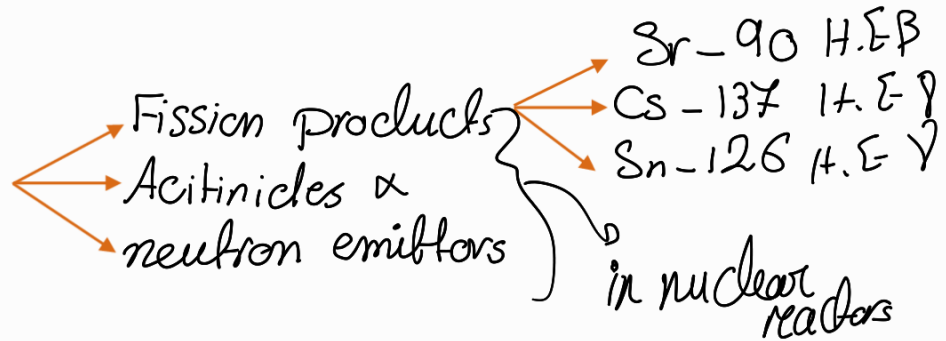
○ Front end

- Fuel preparation

- Main by-product of enrichment is DU (U-238 & U-235) → Used in keels of yachts & anti-tank shells

○ Back end

- Spent fuel rods



- Management: By destroying, diluting or isolating radioactive element

1. Fuel reprocessing

Waste fuel is temporarily stored then it is processed and used.

solidification → **Vitrification**: Mixing high-level waste with sugar then calcination (Slide 90, 91)

2. Fuel Storing

Stored for several years under water in cooling ponds to provide cooling and radiation protection

3. Burials: Under the sea or under the ground

EPA limits

SO_2 : Coal: $\frac{0.52}{10^6} \frac{kg}{KJ}$ / In 1978 stand.
Oil: $\frac{0.34}{10^6} \frac{kg}{KJ}$ 90% reduction
For High Sulfur
Coal

NO_x : Coal: $\frac{0.3}{10^6} \frac{kg}{KJ}$ 70% reduction
For low Sulfur
Coal
Oil: $\frac{0.13}{10^6} \frac{kg}{KJ}$

gas: $\frac{0.086}{10^6} \frac{kg}{KJ}$

Particulates: Coal } $0.043 \frac{kg}{KJ}$
gas }
oil }